

SOLUB

Solchoose: solubility products in water (1s)	<div><div>RAD CHOM</div><div>select compound: Ksol</div><div>7: Ag3AsO4→3Ag++AsO4³⁻</div><div>6: AgBrO3→Ag++BrO3⁻: .00</div><div>5: AgBr→Ag++Br⁻: 5.35E-1</div><div>4: AgC2H3O2→Ag++C2H3O2⁻</div><div>3: AgC7H5O2→Ag++C7H5O2⁻</div><div>2: AgCl→Ag++Cl⁻: 5.97E-1</div><div>1: Ag2CO3→2Ag++CO3²⁻: 8.</div><div>0: Ag2C2O4→2Ag++C2O4²⁻</div><div><div>CANCEL OK</div></div></div>	<div><div>RAD CHOM</div><div>select compound: Ksol</div><div>7: PbBr2→Pb2++2Br⁻: .000</div><div>6: Pb(BrO3)2→Pb2++2BrO3⁻</div><div>5: PbCO3→Pb2++CO3²⁻: 1.4</div><div>4: PbCl2→Pb2++2Cl⁻: .000</div><div>3: PbCrO4→Pb2++CrO4²⁻: 2</div><div>2: PbF2→Pb2++2F⁻: .00000</div><div>1: PbO→Pb2++O2²⁻: 1.2E-15</div><div>0: Pb(OH)2→Pb2++2OH⁻: 1.</div><div><div>CANCEL OK</div></div></div>
Solchoose: choose solubility products (1s)	<div><div>RAD CHOM</div><div>select compound: Ksol</div><div>7: Ca3(AsO4)2→3Ca2++2AsO</div><div>6: CaCO3→Ca2++CO3²⁻: .00</div><div>5: CaCrO4→Ca2++CrO4²⁻</div><div>4: CaF2→Ca2++2F⁻: .00000</div><div>3: CaHPO4→Ca2++HPO4²⁻</div><div>2: Ca(OH)2→Ca2++2OH⁻: .0</div><div>1: Ca(IO3)2→Ca2++2IO3⁻</div><div>0: Ca3(PO4)2→3Ca2++2PO4³⁻</div><div><div>CANCEL OK</div></div></div>	<div>SOLUBILITY PRODUCTS OF COMPOUNDS IN WATER MOSTLY AT 25°C</div> <div>Solchoose CHOOSE COMPOUND</div> <div>Ceconc CEQ + 'R=CRS/CLS'</div> <div>EQ WITH CONCENTRATIONS</div> <div>Molwt ANBN + MOLWEIGHT</div> <div>Cexmol CEQ ↔ Cnd A1.3Cnd B13</div> <div>CL/R SIDE MOLECULES</div> <div>Lsolpr 2 + 3 LIST WITH</div> <div>SOLUBILITY PRODUCTS</div> <div><div>GRAPH</div><div><div>CANCEL OK</div></div></div>
Help to solubility products		
Solchoose: solubility product of AgBr (1s)	<div><div>RAD XYZ DEC R= 'X'</div><div>CHOME SOLUB3 USR</div><div>7:</div><div>6:</div><div>5:</div><div>4:</div><div>3:</div><div>2:</div><div>1: "AgBr→Ag++Br⁻"</div><div>Ksol:5.35E-13</div><div>Solch Cecon Molwt Cexmol Lsolpr Helps</div></div>	<div><div>RAD XYZ DEC R= 'X'</div><div>CHOME SOLUB3 USR</div><div>7:</div><div>6:</div><div>5:</div><div>4:</div><div>3:</div><div>2:</div><div>1: "AgNO2→Ag++NO2⁻"</div><div>Ksol:.0006</div><div>Solch Cecon Molwt Cexmol Lsolpr Helps</div></div>
of AgNO2 (1s)		
Ceconc: equation with concentrations (1s)	<div><div>RAD XYZ DEC R= 'X'</div><div>CHOME SOLUB3 USR</div><div>5:</div><div>4:</div><div>3:</div><div>2:</div><div>1: "AgBr→Ag++Br⁻"</div><div>$K = \frac{c_{Ag^+} \cdot c_{Br^-}}{c_{AgBr}}$</div><div>Solch Cecon Molwt Cexmol Lsolpr Helps</div></div>	<div><div>RAD XYZ DEC R= 'X'</div><div>CHOME SOLUB3 USR</div><div>4:</div><div>3:</div><div>2:</div><div>1: "3Fe3O4+8Al→9Fe+4Al₂O₃"</div><div>$K = \frac{c_{Fe}^9 \cdot c_{Al_2O_3}^4}{c_{Fe_3O_4}^3 \cdot c_{Al}^8}$</div><div>Solch Cecon Molwt Cexmol Lsolpr Helps</div></div>